NEW STANDARD ACADEMY

Semri Kothi Super Market, Raebareli CLASS 11 DPP (Academy) 14-07-2025

PHYSICS

- 1. Give an example of a body moving with uniform speed but having a variable velocity and an acceleration which remains constant in magnitude but changes in direction
- **2.** What will be the effect on horizontal range of a projectile when its initial velocity is doubled, keeping the angle of projection same?
- **3.** What will be the effect on the maximum height of a projectile when its angle of projection is changed from 30° to 60°, keeping the same initial velocity of projection?
- **4.** What is the angular velocity of the hour hand of a clock?
- **5.** A body is moving on a curved path with a constant speed. What is the nature of its acceleration?
- **6.** State, for each of the following physical quantities, if it is a scalar or a vector: volume, mass, speed, acceleration, density, number of moles, velocity, angular frequency, displacement, angular velocity.
- 7. Pick out the two scalar quantities in the following list: force, angular momentum, work, current, linear momentum, electric field, average velocity, magnetic moment, relative velocity.
- **8.** What is the angle between two forces of 2N and 3N having resultant as 4N?
- **9.** At what range will a radar set show a fighter plane flying at 3km above its centre and at a distance of 4km from it?
- **10.** Two forces 5kg wt and 10kg wt are acting with an inclination of 120° between them. What is the angle which the resultant makes with 10kg wt?

CHEMISTRY

- 1. Calculate the valume occupied by 10^{22} molecules of N₂at 27°C and one atmospheric pressure.
- 2. What mass (in kilogram) of K₂O contains the same number of moles of K atoms as are present in one kg KCI?
- 3. Calculate the number of molecules present in 12.3 g MgSO₄.7H₂O. Calculate the mass of Na₂CO₃ which will have molecules equal to those present in 12.3g of MgSO₄.7H₂O.

- 4. How many quantum number are obtained from schrodinger wave equation ? which quantum number is not obtained from this equation?
- 5. What is the ratio of wavelengths associated with N_2 and CO moving with the same velocity?
- 6. The electronic energies in an atom are quantized . Results of which experiment support it?
- 7. Calculate $\frac{e}{m}$ ratio for electron
- 8. Decide the position of an element having atomic number 17 in periodic table.
- 9. Decide the position of an element having atomic number 19
- 10. Write the electronic configuration of the element Z=58 and decide its period, group and block in the periodic table.

BIOLOGY

- 1. Describe the various types of placentation found in flowering plants.
- 2. How do the various leaf modification help plants?
- 3. Describe the arrangement of floral members in relation to their insertion on thalamus.
- 4. Draw the labell diagram -male reproductive system of frog
- 5. Draw the labell diagram of female reproductive system of frog
- 6. What is the difference between monocot and dicot stem
- 7. Give the difference between raceemose and cymose inflorescence
- 8. What is inflorescence
- 9. Define hypogynous flower with example
- 10. Give the difference between simpl and compound leaf

MATHS

- 1. Find the angle in radian between the hands of a clock at 3.30 a.m.
- 2. The difference between two acute angles of a right angled triangle is $\frac{\pi}{5}$ in radian measure. Find these angle in degrees.
- 3. The angles of a triangle are in A.P. and the greatest angle is double the least. Find all the angles in circular measure.
- 4. If $\tan x = -\frac{4}{3}$, find the value of $9 \sec^2 x 4 \cot x$.
- 5. If cosec $x \cot x = \frac{3}{2}$, find $\cos x$. In which quadrant does x lie?
- 6. Find the values of the following:

(i)
$$\cot\left(-\frac{7\pi}{4}\right)$$
 (ii) $\sin\left(-\frac{17\pi}{3}\right)$

7. Find the value of : (i) cos 495° (ii) cot(-315°)

- 8. Prove that $:\sin(690^\circ)\cos(300^\circ) + \cos(-750^\circ)\sin(-240^\circ) = 1$
- 9. Using t-ratios of 30° and 45°, evaluate
 - (i) sin 15° (ii) tan 15° (iii) cos 75°
- 10. Evaluate $\tan \frac{13\pi}{12}$.